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orbital? Answer: Azimuthal quantum number. 2) What is the maximum capacity of d-orbital? Answer: 10 electrons.

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structure is the body diagonal of the cube. (CC BY-NC-SA; anonymous by request) Table 12.1 compares the packing efficiency and the number of nearest neighbors for the different cubic and close-packed structures; the number of nearest neighbors is called the coordination ...

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mixture (solution), or heterogeneous mixture: a) orange juice b) brass c) 0.9% saline (NaCl) solution (freshly-squeezed) d) garden soil e) room air f) methane gas g) sodium metal h) N<sub>2</sub> gas i) Cu(NO<sub>3</sub>)<sub>2</sub> ...

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Define the first ionization energy.

Answer: The enthalpy change when one electron is removed from one mole of gaseous atoms to form one mole of  $+1$

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charged ions. 2) Define covalent bonds.  
Answer: Electrostatic attraction between  
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For molecules with the same number of carbon atoms and the same general shape, the boiling points usually differ only slightly, just as we would expect for substances whose molar mass differs by only 2 u (equivalent to two hydrogen

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atoms). Like other hydrocarbons, the alkenes are insoluble in water but soluble in organic solvents.

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Lakhmir Singh Class 9 CBSE Chemistry  
Text Book PAGE No 150 Solution 01 Ions  
Solution 02 (a). Anions ; (b). Cations

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Solution 03 The formula mass of an ionic compound is the relative mass of its 'formula unit' as compared with the mass of a Carbon-12 atom taken as 12 units. Lakhmir Singh Class 9 CBSE Chemistry Text Book PAGE NO 151  
Solution 04

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